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## Tetrahedron

# Enantiopure substituted pyridines as promising antimalarial drug candidates 

Guillaume Bentzinger ${ }^{\text {a }}$, Etienne Pair ${ }^{\text {a }}$, Jean Guillon ${ }^{\text {b }}$, Mathieu Marchivie ${ }^{\mathrm{c}}$, Catherine Mullié ${ }^{\text {a }}$, Patrice Agnamey ${ }^{\text {a }}$, Alexandra Dassonville-Klimpt ${ }^{\text {a }}$ and Pascal Sonnet ${ }^{{ }^{\text {a* }}}$<br>${ }^{\text {a }}$ Université de Picardie-Jules-Verne, AGIR - Agents infectieux, résistance et chimiothérapie, UR 4294, UFR de pharmacie, 1, rue des Louvels, Amiens, F-80037 cedex 1, France<br>${ }^{\mathrm{b}}$ Université de Bordeaux, UFR des sciences pharmaceutiques, INSERM U1212/UMR CNRS 5320, Laboratoire ARNA, 146, rue Léo-Saignat, Bordeaux, F-33076 cedex, France<br>${ }^{\text {c}}$ CNRS, Université de Bordeaux, Bordeaux INP, ICMCB, UMR 5026, Pessac, F-33600, France

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#### Abstract

We describe the enantioselective synthesis and biological evaluation of 4-(2-amino-1hydroxyethyl)pyridines (4 AHPs) as new antimalarial drug candidates. In particular, two routes to obtain the key-intermediate 4 -vinyl-pyridine were studied. These routes are based on a Kröhnke-type cyclization or on metal-catalyzed reactions. The Kröhnke-type cyclization route is faster but only efficient at low scale since this pathway involves a Wittig reaction that requires severe temperature-control. Consequently, we designed a second route based on metal-catalyzed reactions. This way is longer but the 4 -vinyl-pyridine can be obtained on a 5 g scale at least. Finally, a regioselective $\mathrm{S}_{\mathrm{N}} 2$ ring-opening of enantiopure epoxides by alkyl primary amines allowed the synthesis of eight 4-AHPs with global yields up to $41 \%$. These compounds show strong in vitro antimalarial activity against $P$. falciparum strains and are more active that chloroquine and mefloquine. These results demonstrate that 4-AHPs are promising antimalarial drug candidates.


[^0]
## 1. Introduction

Malaria is still today a major public health threat on a worldwide scale. In 2018, 228 million cases of malaria were reported worldwide and led to 405,000 deaths, $67 \%$ of them children aged under 5 years. ${ }^{1}$ Since 2014, investments for malaria control and elimination were robust and resolute ( 2.7 billion dollars in 2018) but falling far the 5 billion dollars funding target of the World Health Organization (WHO) global strategy. Plasmodium falciparum is the most prevalent malaria parasite in Africa and is responsible for most malaria cases. It has shown great adaptive potential and the apparition of resistances towards antiplasmodial molecules always calls for new drug candidates. In the fight against malaria, the family of arylaminoalcohol (AAA) drugs including quinine ( QN ), mefloquine $(\mathrm{MQ})$ and lumefantrine play an important role. ${ }^{1}$ Not only do they have good activities against $P$. falciparum but their slow clearance makes them perfect partners for artemisinin-based combination therapy (ACT), widely recommended by the WHO since 2006. ${ }^{2}$ MQartesunate and lumefantrine-artemether are among the six ACT commercially available. ${ }^{1,3}$

The first AAA used in antimalarial therapy was the enantiomerically pure quinine, one of the four cinchona alkaloids. Quinidine (QD), the dextrorotatory diastereomer of quinine, was discarded because it differs from QN in many pharmacological respects. Although, QD is about three to four times more active as an antimalarial drug, ${ }^{4}$ the QT (Q wave-T wave interval) prolongation observed with it is about four times more than with QN. ${ }^{5}$ Accordingly, QN was used as antimalarial drug while QD was for years available as an anti-arrhythmic. For the MQ, another AAA, several studies highlighted that the $(R, S)$ enantiomer of MQ is the least active (Fig. 1) and crosses more easily the blood-brain barrier. ${ }^{6-10}$ This latter ability making it, in part, responsible for the important neurological side effects of MQ. ${ }^{11-13}$ Despite important disparities in activities, toxicities and pharmacokinetic profiles of the AAA stereoisomers, current commercial drugs containing MQ or lumefantrine are still used as racemates.

For those reasons, our group works on the enantioselective synthesis of new AAA analogs. Previously, the synthesis of 4-(2-amino-1-hydroxyethyl)quinolines (4-AHQs) as enantiopure open analogs of MQ was achieved (Fig. 1). Structure-activityrelationship (SAR) studies on the aminoalcohol moiety allowed us to identify a few molecules 1a-d with alkyl side chains, as active as MQ in vitro and in vivo on the selected P. falciparum strains 3D7 and W2. ${ }^{14-18}$ These compounds 1a-d also showed a clearer $(S) /(R)$ selectivity in vitro than MQ with $\mathrm{IC}_{50}$ up to 15 -fold higher for the more active enantiomer.

The importance of the aminoalcohol stereochemistry for the antimalarial activity was reinforced with our study on the pyrroloquinoxaline family. ${ }^{18}$ Although their corresponding activity was low (micromolar range), strong ( $S$ )/(R) disparities have been observed (eudysmic ratio up to 15.5).

[^1]Previous Work: 4-AHQs 1


Lead compounds; 1a-d: $\mathrm{n}=2-5$
$\mathrm{IC}_{50}($ P P 3 D 7$)=12.7-254.0 \mathrm{nM}^{15}$
$\mathrm{IC}_{50}($ PAN2 $)=7.0-142.0 \mathrm{nM}^{15}$ ${ }^{1 C_{50}}(S) \ll \mathrm{IC}_{50}(R): 2-15$ times $^{15}$

Fig. 1. Previous work: mefloquine and 4-AHQs 1a-d.
As a continuation of our SAR studies, we decided now to focus our work on the aromatic ring role by restructuring the ring system. Dissociation of the quinoline core by splitting the benzo compound could lead to new antimalarial drugs such as 4-(2-amino-1-hydroxyethyl)pyridines (4-AHPs) 2a-d. Furthermore, 4-AHPs could also be considered as enpiroline (ENP) analogs (Fig. 2). ENP is an AAA based on a pyridine core and was previously studied as a drug candidate. ${ }^{19-22}$ Surprisingly, this compound was never fully exploited despite promising activities, good pharmacokinetic parameters, and low toxicity. Also, ENP enantiomers have similar activities with eudysmic ratios around 1.2 . ${ }^{6,23}$ We thought that 4-AHPs 2a-d could display improved antimalarial activities and enhanced eudysmic ratios compared to 4-AHQs 1a-d and lower long-term toxicity than MQ. ${ }^{15}$ Herein, we describe the enantioselective synthesis and biological evaluation of 4-AHPs 2a-d as new antimalarial potential drugs.


Fig. 2. Present work: Enpiroline and 4-AHPs 2a-d.
In order to take advantage of the strategy we had optimized for generating the library of enantiopure 4-AHQs $\mathbf{1}$ for making the enantiopure 4-AHPs 2 , we needed to access to the epoxides $\mathbf{3}$ through an indirect asymmetric epoxidation of the 4 -vinylpyridine 4 (scheme 1). ${ }^{14}$ Two routes to obtain the keyintermediate 4 -vinyl-pyridine 4 are studied based on a Kröhnketype cyclization ${ }^{24}$ or metal-catalyzed reactions. The respective advantages and inconveniences of these ones are discussed below. Finally, the 4 -AHPs 2a-d were prepared in three steps from the vinyl pyridine $\mathbf{4}$ and their in vitro activities were measured against $P$. falciparum 3D7 and W2 strains. Thus, novel SAR were highlighted with regard to their stereochemistry, alkyl substitution and in comparison with the corresponding quinolines 4-AHQs 1.

## 2. Results and discussion

### 2.1. Chemistry

Two ways of making the 4 -vinylpyridine 4 were explored either via a Kröhnke-type cyclization (7 steps, $46 \%$ overall yield) $)^{24}$ or metal-catalyzed reactions ( 9 steps, $37 \%$ overall yield).


Scheme 1. Retrosynthesis strategies for the 4-AHPs 2a-d.
2.1.l.Synthesis of 4 using a Kröhnke-type
cyclization

Having a Wittig reaction in mind for the synthesis of the enone 6, the commercial 2-bromo-4'(trifluoromethyl)acetophenone $\mathbf{8}$ was treated with $\mathrm{PPh}_{3}$ (Scheme 2 ). The resulting phosphonium salt $\mathbf{9}$, obtained quantitatively, was then reacted with glyoxylic acid to obtain 6 as the Wittig product in quantitative yield. The ${ }^{1} \mathrm{H}$ NMR analysis and the coupling constant between the ethylenic protons $\left({ }^{3} J_{\mathrm{H} / \mathrm{H} 2}=15.6\right.$ $\mathrm{Hz})$ confirmed the selective formation of the $(E)$-isomer. In parallel, the commercially available 3-bromo-1,1,1trifluoroacetone $\mathbf{1 0}$ was reacted with pyridine in EtOH at $70^{\circ} \mathrm{C}$. The corresponding pyridinium salt $\mathbf{5}$ was afforded in $72 \%$ yield with the ketone function completely hydrated due to its electrowithdrawing environment. ${ }^{19,24,25}$ This structure was confirmed by ${ }^{13} \mathrm{C}$ NMR $\left(\delta \mathrm{C}\left(\mathrm{C}(\mathrm{OH})_{2}\right): 148.2 \mathrm{ppm}\right)$ and X-ray diffraction (Scheme 2, supplementary material). With the enone 6 and the pyridinium salt 5 in hands, a Kröhnke-type cyclization was performed in order to form the isonicotinic acid $\mathbf{1 1}$ in $94 \%$ yield. Thereafter, the acid function is converted into aldehyde in order to introduce the vinyl group with a Wittig reaction. Different reducing agents were tested to achieve the direct reduction of the acid moiety to the aldehyde $\mathbf{1 3}$, but mixtures of the aldehyde $\mathbf{1 3}$ and the alcohol $\mathbf{1 2}$ were afforded systematically. Thus, a two steps aldehyde synthesis was performed: i) borane $\left(\mathrm{BH}_{3}\right)$ was used as reducing agent to obtain the primary alcohol 12 in $96 \%$ yield and ii) Dess-Martin Periodinane as selective oxidant allowed to afford the aldehyde $\mathbf{1 3}$ in $81 \%$ yield. For the Wittig reaction, the methyltriphenylphosphonium bromide was deprotonated in situ with $n-\mathrm{BuLi}$ as the base at $0^{\circ} \mathrm{C}$ in THF to generate the corresponding ylide. Due to the high reactivity of this ylide with aldehydes, the reaction required serious optimization (temperature, additives, reaction times). In our best conditions, the addition of aldehyde $\mathbf{1 3}$ was carried out dropwise at $-78{ }^{\circ} \mathrm{C}$. Then, the Wittig reaction itself was performed at $30^{\circ} \mathrm{C}$ for six hours. Thus, the vinyl derivative $\mathbf{4}$ was obtained in $63 \%$ yield. When one gram or more of aldehyde were used, yields dropped between $15 \%$ and $30 \%$. This reaction being very temperature sensitive, it proved to be hard to reproduce when scaled-up, probably due to a lack of temperature-control. With this synthetic pathway, the desired vinyl intermediate 4 is accessible in a $46 \%$ yield in an overall 7 steps sequence. Because of the scalability issues with the final Wittig step, we considered this method to be unsuitable for making a library of compounds and devised a new route to obtain the key derivative $\mathbf{4}$, based on robust and reliable metal-catalyzed reactions.


Scheme 2. Preparation of the vinyl 4 by Kröhnke-type cyclization. Reagents and conditions: (a) $\mathrm{PPh}_{3}, \mathrm{THF}, 70^{\circ} \mathrm{C}$. (b) Glyoxylic acid, $\mathrm{NEt}_{3}$, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(1 / 1), 25^{\circ} \mathrm{C}$. (c) Pyridine, EtOH, $80^{\circ} \mathrm{C}$. (d) $\mathrm{NH}_{4} \mathrm{OAc}$, THF, $70^{\circ} \mathrm{C}$. (e) $\mathrm{BH}_{3}$.DMS, THF, $25^{\circ} \mathrm{C}$. (f) DMP, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}$. (g) 1) $\mathrm{BrPPh}_{3} \mathrm{CH}_{3}, n$-BuLi, LiOH, THF, $0^{\circ} \mathrm{C}$, 2) $13,-78^{\circ} \mathrm{C}, 1 \mathrm{~h}$ then $-30^{\circ} \mathrm{C}, 6 \mathrm{~h}$.

### 2.1.2. Synthesis of 4 using metal-catalyzed reactions.

The commercially available 2,6-dibromopyridine 7 was converted into 4-hydroxypyridine 14 via the formation of a boronate intermediate using the $[\operatorname{Ir}(\operatorname{cod}) \mathrm{Cl}]_{2}$ catalyst in the presence of pinacolborane and 4,4'-di-tert-butyl-2,2'-dipyridyl (dtbpy). Then, a one-pot oxidative cleavage of the boronate was performed with Oxone ${ }^{\circledR}$ to afford 14 in an $82 \%$ yield. ${ }^{26}$ An $O$ protecting reaction was carried out with 4-methoxybenzyl chloride to give 15. A mono-halogen exchange, using Grignard reagent in the presence of $\mathrm{I}_{2}$, afforded the mono-iodinated compound 16 in $85 \%$ yield. After optimization, the best conditions were found by using 1.6 equivalents of both Turbo Grignard and $\mathrm{I}_{2}{ }^{27}$ The mono-iodination allowed to perform a selective Suzuki coupling with 4-(trifluoromethyl)phenylboronic acid affording 17 in $95 \%$ yield. ${ }^{28}$ At that point, direct conversion of $\mathbf{1 7}$ into $\mathbf{1 9}$ using reaction conditions described by Z . Gonda et al. ${ }^{29}$ was unsuccessful. Consequently, a copper-catalyzed aromatic Finkelstein reaction was used in order to convert 17 into 18 in $88 \%$ yield. ${ }^{30,31}$ The iodo derivative $\mathbf{1 8}$ proved to be more reactive and was converted into a $-\mathrm{CF}_{3}$ affording 19 in a satisfactory $89 \%$ yield. ${ }^{29}$ After removal of the -PMB group, the resulting alcohol 20 was reacted with $\mathrm{POBr}_{3}$ to isolate the bromo compound 21 in $93 \%$ yield. This allowed a Suzuki coupling with potassium (vinyl)trifluoroborate affording the key intermediate 4 in $98 \%$ yield. The global yield for this new synthetic pathway is of $37 \%$ over 9 steps. The selected reactions proved to be very robust whatever the scale (up to 5 g of vinyl $\mathbf{4}$ per sequence).

### 2.1.3. Synthesis of 4-(2-amino-1-hydroxyethyl) pyridines 2 (4-AHPs).

With intermediate $\mathbf{4}$ in hands, the diols $(\boldsymbol{S})-\mathbf{2 2}$ and $(\boldsymbol{R})$-22 were obtained enantioselectively using a Sharpless dihydroxylation in the presence of AD-mix- $\alpha$ or $-\beta$ respectively in $99 \%$ yield and $99 \% e e$. A ring-closure was carried out in order to form the epoxides ( $\boldsymbol{S}$ )-3 and ( $\boldsymbol{R}$ )-3 by means of a slightly modified "onepot" reaction, previously described. ${ }^{14}$ Those were obtained with complete retention of configuration in $95 \%$ and $85 \%$ yield from
$(S)-22$ and $(\boldsymbol{R})-\mathbf{2 2}$ respectively. X-ray diffraction of epoxide (S)-3 was performed and the ( $S$ )-geometry was ascribed unambiguously (Scheme 4, supplementary material).


Scheme 3. Preparation of the vinyl 4 using metal-catalyzed reactions. Reagents and conditions: (a) 1) $[\operatorname{Ir}(\operatorname{cod}) \mathrm{Cl}]_{2}$, pinacolborane, dtbpy, cyclohexane, $80^{\circ} \mathrm{C}$, 2) Oxone, THF/ $\mathrm{H}_{2} \mathrm{O}(1 / 1), 25^{\circ} \mathrm{C}$. (b) 4-methoxybenzyl chloride, $\mathrm{K}_{2} \mathrm{CO}_{3}$, DMF, $110^{\circ} \mathrm{C}$. (c) $i \mathrm{PrMgCl}$.LiCl, $\mathrm{I}_{2}$, THF, $25^{\circ} \mathrm{C}$. (d) $4-$ (trifluoromethyl)phenyl boronic acid, $\mathrm{Na}_{2} \mathrm{CO}_{3}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, toluene $/ \mathrm{H}_{2} \mathrm{O} / \mathrm{EtOH}$ (6/1/1), $110^{\circ} \mathrm{C}$. (e) CuI, NaI, $N, N^{\prime}$-dimethylethylenediamine, 1,4-dioxane, $110^{\circ} \mathrm{C}$. (f) KF, CuI, $\mathrm{TMSCF}_{3}, 1,10$-phenantroline, $\mathrm{B}(\mathrm{OMe})_{3}$, DMSO, $60^{\circ} \mathrm{C}$. (g) TFA, DCM, $25^{\circ} \mathrm{C}$. (h) $\mathrm{POBr}_{3}$, DMF, $110^{\circ} \mathrm{C}$. (i) Potassium (vinyl)trifluoroborate, $\quad \mathrm{Na}_{2} \mathrm{CO}_{3}, \quad \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}, \quad$ toluene/EtOH/H2 $\mathrm{H}_{2} \mathrm{O} \quad(6 / 1 / 1)$, $110^{\circ} \mathrm{C}$.

The last step of the synthesis consisted in a regioselective $S_{N} 2$ ring-opening with diverse primary amines (Scheme 4, Table 1). Using microwave irradiations allowed for reduced reaction times compared to the classical heating that was used for making the corresponding quinoline derivatives. Eight (S)- or (R)-4-AHPs 2a-d were thus synthesized using four alkylamines of growing size in good to excellent yields (60-96\%) and with enantiomeric excesses superior to $98 \%$.


Scheme 4. Synthesis of 4-AHPs 2a-d. Reagents and conditions: (a) AD-mix$\alpha$ or $-\beta, \mathrm{K}_{2} \mathrm{OsO}_{2}(\mathrm{OH})_{4}, t \mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(1 / 1), 0^{\circ} \mathrm{C}$ to $25^{\circ} \mathrm{C}, 15 \mathrm{~h}$. (b) i) trimethyl orthoacetate, $p \mathrm{TsOH}, \mathrm{DCM}, 25^{\circ} \mathrm{C}, 7 \mathrm{~h}$, ii) TMSBr, DCM, $25^{\circ} \mathrm{C}, 15 \mathrm{~h}$, iii) $\mathrm{K}_{2} \mathrm{CO}_{3}$, MeOH, $25^{\circ} \mathrm{C}, 4$ h. (c) $\mathrm{R}^{1}-\mathrm{NH}_{2}$, EtOH, $130^{\circ} \mathrm{C}$ (M.W.), 30 min .

### 2.2. Biological evaluation.

The activities of all 4-AHPs 2a-d were evaluated against $P$. falciparum 3D7 and W2 using the SYBR Green I method. ${ }^{32}$ $P f \mathrm{~W} 2$ is a chloroquine (CQ) resistant strain and is MQ sensitive while Pf3D7 is chloroquine sensitive and displays a decreased susceptibility to MQ. The two antiplasmodial compounds CQ
and MQ were used as references. The half maximal inhibitory concentration $\left(\mathrm{IC}_{50}\right)$ calculated are reported in Table 2.

Table 1.
Yields and enantiomeric purities of the synthesized 4-AHPs 2a-d.

| Entry | Compound | -NHR ${ }^{1}$ | Yield | $\% e e^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | (S)-2a |  | $78 \%$ | 99 |
| 2 | (R)-2a |  | 96\% | 99 |
| 3 | (S)-2b |  | 94\% | 99 |
| 4 | (R)-2b |  | 85\% | 99 |
| 5 | (S)-2c |  | 67\% | 99 |
| 6 | (R)-2c |  | 65\% | 98 |
| 7 | (S)-2d |  | 60\% | 99 |
| 8 | (R)-2d |  | 85\% | 99 |

${ }^{\text {a }}$ determined by chiral HPLC, see experimental part.
Pleasingly, all $\mathrm{IC}_{50}$ are in the nanomolar range and all compounds are more active than the references whatever the strain. As previously observed for the AHQs 1a-d, ${ }^{15}$ the 4-AHPs 2a-d are more active against PfW2 than Pf3D7, the CQ-resistant strain, with $\mathrm{IC}_{50}$ ranging from 3.5 (entry 8) to 10.0 nM (entry 2). For the Pf3D7 strain, less sensitive to MQ , the $\mathrm{IC}_{50}$ are between 17.7 (entry 7) and 56.7 nM (entry 2). In general, it appears that a longer side chain enhances the efficacy of the compound and the differences between $(S)$ - and $(R)$ - enantiomers. Interestingly, this trend correlates with a higher calculated $\log P$ of the corresponding molecules. Compound 2d, with a heptyl side chain, shows the best activities against both Pf3D7 and PfW2 and has a clear difference between $(S)$ - and $(R)$ - enantiomers against Pf3D7 with a eudysmic ratio of 3.1. The newly synthesized 4-AHPs 2a-d are as much active as their quinoline counterparts 4-AHQs 1, but clearly show less differences between $(S)$ - and $(R)$ - enantiomers which had a eudysmic ratio between 2.2 and 15.1. ${ }^{15}$ Nonetheless, disparities could appear more clearly in pharmacokinetic properties like in the case of MQ. These pharmacokinetic studies are currently under progress.

## 3. Conclusion

The synthesis of eight enantiopure AHPs 2a-d with varying alkyl substituents was achieved from the 4 -vinylpyridine 4. This key-intermediate was obtained by two routes using either a Kröhnke-type cyclization or metal-catalyzed reactions. The cyclisation route led to 4 in a $46 \%$ yield in an overall 7 steps sequence. However, this pathway was not suitable for making a library of compounds, as the scaling-up was limited by the last step of the sequence. The metal-catalyzed route proved to be more robust. However, 4 was obtained in a lower yield of $37 \%$ yield over 9 steps. Nonetheless, this new synthesis allowed us to obtain large quantities of the AHPs precursor 4 (up to 5 g per sequence). Consecutively, a regioselective $S_{N} 2$-ring opening of enantiopure epoxides $\mathbf{3}$ with four alkyl primary amines gave 4AHPs 2a-d with good yields and excellent ee. These pyridines 2a-d showed strong antimalarial activity with $\mathrm{IC}_{50}$ ranging from 3.5 to 10.0 nM against $P f 3 \mathrm{D} 7$ and 17.7 to 56.7 nM against $P f \mathrm{~W} 2$. Compared with their quinoline counterparts AHQs 1, the differences between $(S)$ - and $(R)$ - enantiomers $\mathrm{IC}_{50}$ are less marked with eudysmic ratio between 1.1 and 3.1. These novel
promising antimalarial 4-AHPs 2a-d and derivatives were recently patented. ${ }^{32}$ Pharmacokinetic studies and in vivo
experiments in a mouse model are under progress to validate their potential as antimalarial drug candidates.

## Table 2.

In vitro antiplasmodial activities of the 4-AHPs 2a-d and references CQ and MQ against a chloroquine sensitive Plasmodium falciparum clone (3D7) and a chloroquine-resistant Plasmodium falciparum clone (W2), eudysmic ratio and clogP values.

| Entry | Compound code | -NHR ${ }^{1}$ | $\mathrm{IC}_{50} \pm \mathrm{SD}(\mathrm{nM})^{\mathrm{a}, \mathrm{b}}$ |  | eudysmic ratio ${ }^{\text {c }}$ |  | $\operatorname{clog} \mathrm{P}^{\text {d }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | Pf $3 \mathrm{D} 7^{\text {e }}$ | Pf $\mathrm{W} 2^{\text {f }}$ | Pf3D7 | PfW2 |  |
| 1 | (S)-2a |  | $45.8 \pm 2.3$ | $9.0 \pm 0.4$ | 1.2 | 1.1 | 4.74 |
| 2 | (R)-2a |  | $56.7 \pm 2.3$ | $10.0 \pm 0.7$ |  |  |  |
| 3 | (S)-2b |  | $52.4 \pm 1.7$ | $8.6 \pm 1.0$ | 1.3 | 1.2 | 5.26 |
| 4 | (R)-2b |  | $41.8 \pm 5.5$ | $7.2 \pm 0.5$ |  |  |  |
| 5 | (S)-2c | $\stackrel{S}{3}_{S_{H}}$ | $47.1 \pm 13.2$ | N.D. ${ }^{\text {g }}$ | 1.5 | N.D. | 5.81 |
| 6 | (R)-2c |  | $32.3 \pm 2.0$ | $8.3 \pm 0.8$ |  |  |  |
| 7 | (S)-2d | $今{ }_{H}^{s}{ }_{H}+H_{5}$ | $17.7 \pm 4.7$ | $5.6 \pm 0.3$ | 3.1 | 1.6 | 6.08 |
| 8 | (R)-2d |  | $54.4 \pm 3.2$ | $3.5 \pm 0.5$ |  |  |  |
| 9 | Mefloquin | MQ) | $75.9 \pm 3,0$ | $198.8 \pm 27.0$ | - | - | 3.91 |
| 10 | Chloroquin | CQ) | $79.7 \pm 8.5$ | $31.8 \pm 1.0$ | - | - | 4.40 |

${ }^{a}$ in vitro measurements against P. falciparum using the SYBR Green I method, see experimental part. ${ }^{b}$ Results expressed as mean $\pm$ standard deviation. ${ }^{\text {c }}$ ratio between the $\mathrm{IC}_{50}$ of the more active enantiomer and the less active. ${ }^{\mathrm{d}}$ Predicted octanol/water partition coefficient calculated with Maestro Material Sciences 2.9 .011 software. ${ }^{\mathrm{e}}$ P. falciparum strain with decreased sensibility to MQ and sensitive to CQ. ${ }^{\mathrm{f}}$ P. falciparum strain resistant to CQ and sensitive to MQ. ${ }^{\mathrm{g}}$ Not determined

## 4. Experimental

### 4.1. Chemistry

Reactions were monitored by thin-layer chromatography with silica gel $60 \mathrm{~F}_{254}$ pre-coated aluminium plates $(0.25 \mathrm{~mm})$. Visualization was performed under UV light and PMA oxidation. Filtrations were performed on Celite ${ }^{\circledR} 545$. Chromatographic purification of compounds was achieved with 60 silica gel (40$63 \mu \mathrm{~m}$ ). Unless otherwise noted, all reagent-grade chemicals and solvents were used as supplied (analytical or HPLC grade) without prior purification. Melting points were measured on a Stuart SMP3 apparatus with a precision of $+/-1.5^{\circ} \mathrm{C}$ and are uncorrected. Infrared spectra (IR) were recorded on a FT/IR-4200 Jasco with an ATR-Golden gate. Liquids and solids were applied on the Single Reflection Attenuated Total Reflectance (ATR) Accessories. Data are reported in $\mathrm{cm}^{-1}$. Optical rotations were determined with a Jasco P1010 polarimeter with a 10 cm cell. Specific rotations are reported in $10^{-1}$ deg. $\mathrm{cm}^{2} . \mathrm{g}^{-1}$ and concentrations in g per 100 mL . ${ }^{1} \mathrm{H}$ NMR Spectra ( 400 MHz ) and ${ }^{13} \mathrm{C}$ NMR spectra ( 100 MHz ) were recorded on a Bruker 400 MHz NMR. The field was locked by external referencing to the relevant deuteron resonance. Data appear in the following order: chemical shifts in ppm which were referenced to the internal solvent signal, number of protons, multiplicity ( $s$, singlet; $d$, doublet; $t$, triplet; $d d$, doublet of doublet, $d d d$, doublet of doublet of doublet, $d d t$, doublet of doublet of triplet, $m$, multiplet) and coupling constant $J$ in Hertz. The abbreviation Ar is used to denote aromatic, $b r$. to denote broad and $a p p$. to denote apparent. Coupling constants, $J$, are measured to the nearest 0.1 Hz and are presented as observed. LC-HRMS analyses were performed on an ACQUITY UPLC H-Class system (WatersMicromass, Manchester, UK) coupled with a SYNAPT G2-Si QTOF hybrid quadrupole time-of-flight instrument (WatersMicromass, Manchester, UK), equipped with an electrospray (ESI) ionization source (Z-spray) and an additional sprayer for the reference compound (Lock Spray)., Torrance, CA, USA) heated at $50{ }^{\circ} \mathrm{C}$. High-resolution mass spectra (HRMS) were obtained from a Micromass-Waters Q-TOF Ultima spectrometer, in electrospray ionization (ESI) mode (positive or negative). Enantiomeric excesses were measured with Schimadzu LC20AD equipped with a Chiralpak column (IA, IB, IC, ID or IG).

### 4.1.1. (2-Oxo-2-(4-(trifluoromethyl)phenyl)ethyl) triphenyl phosphonium bromide (9)

To a solution of 2-bromo-1-(4-(trifluoromethyl)phenyl)ethanone 8 ( $1.00 \mathrm{~g}, 3.70 \mathrm{mmol}, 1 \mathrm{eq}$.) in THF ( 20 mL ) were added triphenylphosphine ( $1.08 \mathrm{~g}, 4.10 \mathrm{mmol}, 1.1 \mathrm{eq}$.$) . The$ reactional mixture was heated to reflux for 15 h and concentrated under reduced pressure. The residue was washed with $\mathrm{Et}_{2} \mathrm{O}$. The solid was filtered, washed with toluene and then with $\mathrm{Et}_{2} \mathrm{O}$ to afford $9\left(2.00 \mathrm{~g}\right.$, quant.) as a white solid. m.p. $136{ }^{\circ} \mathrm{C}$; NMR ${ }^{1} \mathrm{H}$ ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H} 6.41(\mathrm{~d}, J=12.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.49-7.57 (m, $11 \mathrm{H}), 7.84-7.93(\mathrm{~m}, 6 \mathrm{H}), 8.49(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm}$; NMR ${ }^{13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{C} 38.7(\mathrm{~d}, J=61.3 \mathrm{~Hz}), 118.4(\mathrm{~d}$, $J=89.3 \mathrm{~Hz}), 123.3(\mathrm{q}, J=272.9 \mathrm{~Hz}), 125.8(\mathrm{q}, J=3.6 \mathrm{~Hz})$, 130.1 (d, $J=13.1 \mathrm{~Hz}), 130.4,133.4(\mathrm{~d}, J=10.7 \mathrm{~Hz}), 134.8(\mathrm{~d}, J$ $=2.9 \mathrm{~Hz}), 135.4(\mathrm{q}, J=32.7 \mathrm{~Hz}) 137.6(\mathrm{~d}, J=5.6 \mathrm{~Hz}), 191.2(\mathrm{~d}$, $J=10.9 \mathrm{~Hz}) \mathrm{ppm}$; IR $v_{\text {max }}: 3435,3057,2722,1680,1409,1316$, $1107 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{27} \mathrm{H}_{21} \mathrm{~F}_{3} \mathrm{OP}[\mathrm{M}+\mathrm{H}]^{+}$449.1282, found 449.1285 .

### 4.1.2. (E)-4-oxo-4-(4-(trifluoromethyl)phenyl)but-

 2-enoüc acid (6)To a solution of 6.00 g ( $11.4 \mathrm{mmol}, 1 \mathrm{eq}$.$) of \mathbf{9}$ in 30 mL of $\mathrm{CHCl}_{3}$ were added $2.00 \mathrm{~mL}\left(1.45 \mathrm{~g}, 17.0 \mathrm{mmol}, 1.5 \mathrm{eq}\right.$.) of $\mathrm{NEt}_{3}$. The solution was stirred at $25^{\circ} \mathrm{C}$ for 5 min , before to add 1.57 g ( $17.0 \mathrm{mmol}, 1.5 \mathrm{eq}$.) of glyoxylic acid in 30 mL of MeOH . The
reactional mixture was stirred at $25^{\circ} \mathrm{C}$ for 15 h and then concentrated in vacuo. AcOEt ( 30 mL ) were added before to be treated with $4 \times 30 \mathrm{~mL}$ sat. aq. $\mathrm{NaHCO}_{3}$. The aqueous layers were combined and acidified ( $\mathrm{pH} \sim 1$ ) with 1 M aq . HCl . The solid was filtered and dried to afford $\mathbf{6}(3.07 \mathrm{~g}$, quant.) as a white solid. m.p. $154{ }^{\circ} \mathrm{C}$; NMR ${ }^{1} \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta \mathrm{H} 6.80(\mathrm{~d}, J=$ $15.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.82(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.90(\mathrm{~d}, J=15.6 \mathrm{~Hz}, 1 \mathrm{H})$, $8.15(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ; \operatorname{NMR}{ }^{13} \mathrm{C}\left(75 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta \mathrm{C}$ $129.4(\mathrm{q}, J=272.0 \mathrm{~Hz}), 127.0(\mathrm{q}, J=3.8 \mathrm{~Hz}), 130.5,134.8$, 135.7 (q, $J=32.5 \mathrm{~Hz}$ ), 137.0, $140.9,168.2,190.3 \mathrm{ppm}$; IR $v_{\text {max }}$ : 2982, 1691, 1665, 1504, 1170, 1134, $1111 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~F}_{3} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 267.0245$, found 267.0238 .

### 4.1.3. 1-(3,3,3-Trifluoro-2,2-dihydroxypropyl) pyridin-1-ium bromide (5)

To a solution of 5.00 g ( $26.3 \mathrm{mmol}, 1.3 \mathrm{eq}$.) of 3-bromo-1,1,1-trifluoropropan-2-one $\mathbf{1 0}$ in 20 mL of EtOH were added 1.60 mL ( $20.2 \mathrm{mmol}, 1 \mathrm{eq}$.) of pyridine. The mixture was heated to reflux for 10 h , before to be cooled to $25^{\circ} \mathrm{C} . \mathrm{Et}_{2} \mathrm{O}$ was then added until to obtain a murky solution. The mixture was stirred at $25^{\circ} \mathrm{C}$ for 15 h . The solid was then filtered to afford $5(6.35 \mathrm{~g}, 72 \%)$ as a white solid. m.p. $193{ }^{\circ} \mathrm{C}$; NMR ${ }^{1} \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right): \delta \mathrm{H} 1.19$ (t, $J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 3.62(\mathrm{q}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.99(\mathrm{~s}, 2 \mathrm{H}), 8.18(\mathrm{t}, J$ $=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 8.71(\mathrm{t}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.91(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H})$ ppm ; NMR ${ }^{13} \mathrm{C}\left(75 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right): \delta \mathrm{C} 18.4,58.3,63.8,94.9(\mathrm{q}, J=$ $32.0 \mathrm{~Hz}), 125.0(\mathrm{q}, J=291.1 \mathrm{~Hz}), 128.9,148.2 \mathrm{ppm}$; IR $v_{\text {max }}$ : 3157, 1300, 1239, 1152, $1087 \mathrm{~cm}^{-1}$.

### 4.1.4. 2-(Trifluoromethyl)-6-(4-(trifluoromethyl) phenyl) isonicotinic acid (11)

In 300 mL of MeOH were added 2.92 g ( $9.24 \mathrm{mmol}, 1 \mathrm{eq}$.) of $\mathbf{5}, 3.38 \mathrm{~g}$ ( $13.9 \mathrm{mmol}, 1.5 \mathrm{eq}$.) of $\mathbf{6}$ and 5.70 g ( $73.9 \mathrm{mmol}, 8 \mathrm{eq}$.) of $\mathrm{NH}_{4} \mathrm{OAc}$. The reaction mixture was heated to reflux for 24 h before to concentrate it in vacuo. The residue was directly purified by flash chromatography (cyclohexane/ $\mathrm{AcOEt} / \mathrm{AcOH}$ $1 / 1 / 0.01)$ to afford $\mathbf{1 1}(2.92 \mathrm{~g}, 94 \%)$ as an orange solid. m.p. $205{ }^{\circ} \mathrm{C}$; NMR ${ }^{1} \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta \mathrm{H} 7.69(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $2 \mathrm{H}), 8.10(\mathrm{~s}, 1 \mathrm{H}), 8.16(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.46(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm}$; NMR ${ }^{13} \mathrm{C}\left(75 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta \mathrm{C} 120.0,122.4(\mathrm{q}, J=273.6 \mathrm{~Hz})$, $123.9,125.6(\mathrm{q}, J=271.6 \mathrm{~Hz}), 126.8,128.6,132.9(\mathrm{q}, J=$ $32.4 \mathrm{~Hz}), 141.5,143.1,150.5(\mathrm{q}, J=35.2 \mathrm{~Hz}), 158.2,166.3 \mathrm{ppm}$; IR $v_{\text {max }}$ : 2980, 1714, $1128 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{14} \mathrm{H}_{8} \mathrm{~F}_{6} \mathrm{NO}_{2}$ $[\mathrm{M}+\mathrm{H}]^{+} 336.0459$, found 336.0466 .
4.1.5. (2-(Trifluoromethyl)-6-(4-(trifluoromethyl)
phenyl)pyridin-4-yl)methanol (12)

To a solution, at $0^{\circ} \mathrm{C}$ and under argon, of 78.0 mg ( $0.23 \mathrm{mmol}, 1$ eq.) of $\mathbf{1 1} \mathrm{in} 2 \mathrm{~mL}$ of anhydrous THF were added 0.58 mL ( $1.15 \mathrm{mmol}, 5 \mathrm{eq}$.) of $\mathrm{BH}_{3}$.DMS ( 2 M in THF). The solution was stirred at $25^{\circ} \mathrm{C}$ for 8 h , cooled to $0^{\circ} \mathrm{C}$ and treated with an of excess MeOH . The reactional mixture was stirred for 5 min at $0^{\circ} \mathrm{C}$ and 5 min at $25^{\circ} \mathrm{C}$, before concentrated in vacuo. The residue was extracted with $\mathrm{Et}_{2} \mathrm{O}$. The organic layers were combined, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuo. The residue was purified by flash chromatography (cyclohexane/AcOEt 2/1) to afford $\mathbf{1 2}(71 \mathrm{mg}, 96 \%)$ as a light orange solid. m.p. $83{ }^{\circ} \mathrm{C}$; NMR ${ }^{1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{H} 4.90$ $(\mathrm{s}, 2 \mathrm{H}), 7.66(\mathrm{~s}, 1 \mathrm{H}), 7.74(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.95(\mathrm{~s}, 1 \mathrm{H}), 8.18$ (d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm}$; NMR ${ }^{13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{C} 63.1$, $116.8(\mathrm{q}, J=2.7 \mathrm{~Hz}), 120.2,121.5(\mathrm{q}, J=274.5 \mathrm{~Hz}), 124.0(\mathrm{q}, J$ $=272.2 \mathrm{~Hz}), 125.6(\mathrm{q}, J=3.8 \mathrm{~Hz}), 127.5,131.6(\mathrm{q}, J=32.6 \mathrm{~Hz})$, 141.0, $148.6(\mathrm{q}, J=34.7 \mathrm{~Hz}), 153.0,156.4 \mathrm{ppm}$; IR $v_{\max }: 3249$, 2918, 2850, 1323, $1116 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{14} \mathrm{H}_{9} \mathrm{~F}_{6} \mathrm{NONa}$ $[\mathrm{M}+\mathrm{Na}]^{+} 344.0486$, found 344.0496 .
4.1.6. 2-(Trifluoromethyl)-6-(4-(trifluoromethyl)

## phenyl) isonicotinaldehyde (13)

To a solution, at $0{ }^{\circ} \mathrm{C}$, of $90.0 \mathrm{mg}(0.28 \mathrm{mmol}, 1 \mathrm{eq}$.) of $\mathbf{1 2}$ in 2 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ were added $132 \mathrm{mg}(0.31 \mathrm{mmol}, 1.1 \mathrm{eq}$.$) of$ Dess-Martin Periodinane. The suspension was stirred at $0^{\circ} \mathrm{C}$ for 1.5 h before to be treated with sat. aq. $\mathrm{NaHCO}_{3}$. The reaction mixture was stirred at $25^{\circ} \mathrm{C}$ for 5 min . The solid was then filtered and the filtrate extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layers were combined, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuo. The residue was purified by flash chromatography (cyclohexane/AcOEt 1/1) to afford $\mathbf{1 3}(72.0 \mathrm{mg}, 81 \%)$ as a white solid. m.p. $69{ }^{\circ} \mathrm{C}$; NMR ${ }^{1} \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta \mathrm{H} 7.77(\mathrm{~d}, J=$ $8.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.06(\mathrm{~s}, 1 \mathrm{H}), 8.24(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.35(\mathrm{~s}, 1 \mathrm{H})$, $10.21(\mathrm{~s}, 1 \mathrm{H}) \mathrm{ppm}$; NMR ${ }^{13} \mathrm{C}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta \mathrm{C} 118.0,120.2$ $(\mathrm{q}, J=272.0 \mathrm{~Hz}), 121.6,123.9(\mathrm{q}, J=274.5 \mathrm{~Hz}), 126.0,127.6$, $132.1(\mathrm{q}, J=32.7 \mathrm{~Hz}), 139.8,144.2,150.0(\mathrm{q}, J=35.9 \mathrm{~Hz})$, $158.0,189.6 \mathrm{ppm}$; IR $v_{\text {max }}: 1711,1319,1118 \mathrm{~cm}^{-1}$.
4.1.7. 2-(Trifluoromethyl)-6-(4-(trifluoromethyl) phenyl)-4-vinylpyridine (4)

Method A: Wittig reaction: To a suspension, at $0{ }^{\circ} \mathrm{C}$ and under argon, of $729 \mathrm{mg} \quad$ ( $2.04 \mathrm{mmol}, \quad 1.3 \mathrm{eq}$.$) \quad of$ methyltriphenylphosphonium bromide, 75.0 mg ( 3.14 mmol , 2 eq.) of LiOH in 30 mL of distilled THF were added 1.96 mL ( $3.14 \mathrm{mmol}, 2$ eq.) of $n$-BuLi ( 1.6 M in hexane). The reaction mixture was stirred at $0^{\circ} \mathrm{C}$ for 10 min , at $25^{\circ} \mathrm{C}$ for 10 min and cooled to $-78^{\circ} \mathrm{C}$. A solution of 500 mg ( $1.57 \mathrm{mmol}, 1 \mathrm{eq}$.) of $\mathbf{1 3}$ in 10 mL of distilled THF were added dropwise. The solution was stirred for 20 min at $-78^{\circ} \mathrm{C}$ before to be heated to $-30^{\circ} \mathrm{C}$. The reaction mixture was stirred at $-30^{\circ} \mathrm{C}$ for 6 h , heated to $25^{\circ} \mathrm{C}$ in $5 \mathrm{~h}\left(0.2^{\circ} \mathrm{C} / \mathrm{min}\right)$ and stirred at $25^{\circ} \mathrm{C}$ for 15 h . The solution was treated with a MeOH excess before to be extracted with AcOEt. The organic layers were combined, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuo. The residue was purified by flash chromatography (cyclohexane/AcOEt 10/1) to afford 4 ( $270 \mathrm{mg}, 63 \%$ ) as an oil, solidifying over time into a light yellow solid.

Method B: Suzuki coupling: An argon-purged mixture of 21 $(6.0 \mathrm{~g}, 16.3 \mathrm{mmol}, 1 \mathrm{eq}$.$) , potassium vinyltrifluoroborate (2.4 \mathrm{~g}$, $17.9 \mathrm{mmol}, 1.1 \mathrm{eq}.), \mathrm{Na}_{2} \mathrm{CO}_{3}(3.5 \mathrm{~g}, 32.6 \mathrm{mmol}, 2 \mathrm{eq}$.$) and$ $\operatorname{Pd}\left(\mathrm{PPh}_{3}\right)_{4} \quad(0.94 \mathrm{~g}, \quad 0.81 \mathrm{mmol}, \quad 0.05$ eq. $)$ in a mixture of toluene $/ \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(6 / 1 / 1,280 \mathrm{~mL})$ was stirred at $110^{\circ} \mathrm{C}$ for 16 h . The reaction mixture was cooled to $25^{\circ} \mathrm{C}$ and extracted with AcOEt. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 10/1) affording $4(5.08 \mathrm{~g}, 98 \%)$ as a yellow oil, solidifying over time into light yellow solid. m.p. $37{ }^{\circ} \mathrm{C}$; NMR ${ }^{1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{H} 5.66(\mathrm{~d}, J=10.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.15$ (d, $J=17.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.78$ (dd, $J=17.6 \mathrm{~Hz}, 10.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.65(\mathrm{~s}$, $1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.83(\mathrm{~s}, 1 \mathrm{H}), 8.15(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, 2H) ppm; NMR ${ }^{13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta \mathrm{C} 116.2(\mathrm{q}, J=2.7 \mathrm{~Hz})$, $120.3,120.7,121.3(\mathrm{q}, J=272.2 \mathrm{~Hz}), 123.9(\mathrm{q}, J=272.2 \mathrm{~Hz})$, $125.7(\mathrm{q}, J=3.4 \mathrm{~Hz}), 127.4,131.5(\mathrm{q}, J=32.5 \mathrm{~Hz}), 133.7,141.0$, 147.7, 148.9 (q, $J=34.5 \mathrm{~Hz}$ ), 156.7 ppm ; IR $v_{\text {max }}: 1321,1108$ $\mathrm{cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{~F}_{6} \mathrm{~N}[\mathrm{M}+\mathrm{H}]^{+}$318.0717, found 318.0725.

### 4.1.8. 2,6-Dibromopyridin-4-ol (14)

An argon-purged mixture of 2,6-dibromopyridine $7(2.0 \mathrm{~g}$, $8.44 \mathrm{mmol}, 1 \mathrm{eq}$. ), pinacolborane ( $2.20 \mathrm{~mL}, 15.2 \mathrm{mmol}, 1.8 \mathrm{eq}$. ), $[\mathrm{Ir}(\mathrm{cod}) \mathrm{Cl}]_{2}(56.7 \mathrm{mg}, 0.08 \mathrm{mmol}, 0.01 \mathrm{eq}$.$) and 4,4^{\prime}$-di-tert-butyl-$2,2^{\prime}$-dipyridyl ( $45.3 \mathrm{mg}, 0.017 \mathrm{mmol}, 0.02 \mathrm{eq}$.) in cyclohexane $(15 \mathrm{~mL})$ was stirred at $80^{\circ} \mathrm{C}$ for 18 h . The reaction mixture was cooled back to $25^{\circ} \mathrm{C}$ and concentrated under reduced pressure. The residue was dissolved in THF ( 20 mL ) before adding
dropwise a solution of oxone monopersulfate $(2.9 \mathrm{~g}, 9.3 \mathrm{mmol}$, 1.1 eq.) in $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{~mL})$. The resulting mixture was stirred at $25^{\circ} \mathrm{C}$ for 30 min , quenched with $1 \mathrm{M} \mathrm{NaHCO}_{3}$ aq. and extracted with EtOAc. The combined organic layers were washed with water, brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The resulting solid was purified by flash chromatography on silica gel (cyclohexane/EtOAc 4/1) affording $14(1.75 \mathrm{~g}, 82 \%)$ as a white solid. m.p. $207^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta \mathrm{H} 6.97$ (s, 2H, $\mathrm{H}_{3}, \mathrm{H}_{5}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta \mathrm{C} 116.0\left(\mathrm{C}_{3}, \mathrm{C}_{5}\right), 141.8\left(\mathrm{C}_{2}, \mathrm{C}_{6}\right), 168.6$ $\left(\mathrm{C}_{4}\right) \mathrm{ppm}$; IR $v_{\text {max }}: 2838,1540,1147,769 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{5} \mathrm{H}_{4} \mathrm{NO}^{79} \mathrm{Br}_{2}[\mathrm{M}+\mathrm{H}]^{+} 251.8660$, found 251.8663 .

### 4.1.9. 2,6-Dibromo-4-((4-methoxybenzyl)oxy) pyridine (15)

To a solution of $\mathbf{1 4}(8.0 \mathrm{~g}, 31.6 \mathrm{mmol}, 1 \mathrm{eq}$.$) in DMF$ $(100 \mathrm{~mL})$ were added 4-methoxybenzyl chloride ( 4.3 mL , $31.6 \mathrm{mmol}, 1 \mathrm{eq}$.) and $\mathrm{K}_{2} \mathrm{CO}_{3}(5.7 \mathrm{~g}, 41.1 \mathrm{mmol}, 1.3 \mathrm{eq}$.). The suspension was stirred at $110^{\circ} \mathrm{C}$ for 15 h , cooled back to $25^{\circ} \mathrm{C}$, filtered over a celite pad and the filtrate was concentrated under reduced pressure. The residue was placed in EtOAc and washed with 0.5 M NaOH . The aqueous layer was extracted with EtOAc. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 9/1) affording 15 ( $11.7 \mathrm{~g}, 99 \%$ ) as a white solid. m.p. $69{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H} 3.83$ (s, 3 H ), $5.00(\mathrm{~s}, 2 \mathrm{H}), 6.94\left(\mathrm{~d},{ }^{3} J=8.7 \mathrm{~Hz}, 2 \mathrm{H}\right), 7.04(\mathrm{~s}, 2 \mathrm{H}), 7.31\left(\mathrm{~d},{ }^{3} \mathrm{~J}=\right.$ $8.7 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{C} 55.3,70.8$, $114.0,114.2,126.3,129.5,141.1,160.0,166.7 \mathrm{ppm}$; IR $v_{\text {max }}$ : 2925, 1572, 1514, 982, $826 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{13} \mathrm{H}_{11}{ }^{79} \mathrm{Br}_{2} \mathrm{NO}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 395.9054$, found 395.9065.

### 4.1.10. 2-Bromo-6-iodo-4-((4-methoxybenzyl)oxy) pyridine (16)

To a solution of $\mathbf{1 5}(4.97 \mathrm{~g}, 13.1 \mathrm{mmol}, 1 \mathrm{eq}$.$) in anhydrous$ THF ( 13 mL ) were added $i \mathrm{PrMgCl} . \mathrm{LiCl}$ ( 1.3 M in THF) $(16.2 \mathrm{~mL}, 21.0 \mathrm{mmol}, 1.6 \mathrm{eq}$.$) . The reaction mixture was stirred$ at $25^{\circ} \mathrm{C}$ for 2 h and cooled to $0^{\circ} \mathrm{C}$ before adding portionwise $\mathrm{I}_{2}$ $(5.33 \mathrm{~g}, 21.0 \mathrm{mmol}, 1.6 \mathrm{eq}$.). The resulting mixture was stirred at $0^{\circ} \mathrm{C}$ for 5 min , at $25^{\circ} \mathrm{C}$ for 15 h and quenched with 0.5 M $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ aq. ( $3 \times 40 \mathrm{~mL}$ ). The aqueous layer was extracted with EtOAc. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 10/1) affording 16 ( $4.68 \mathrm{~g}, 85 \%$ ) as a white solid. m.p. $96{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{H}$ $3.83(\mathrm{~s}, 3 \mathrm{H}), 4.99(\mathrm{~s}, 2 \mathrm{H}), 6.93(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.04(\mathrm{~d}, J=$ $1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.26-7.34(\mathrm{~m}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta \mathrm{C} 55.3,70.6,114.3,114.3,115.5,121.2,126.4,129.5$, $141.0,160.0,165.6 \mathrm{ppm}$; IR $v_{\max }: 2928,1563,1516,1251,983$, $826 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{13} \mathrm{H}_{12}{ }^{79} \mathrm{BrINO}_{2}[\mathrm{M}+\mathrm{H}]^{+} 419.9096$, found 419.9097.

### 4.1.11. 2-Bromo-4-((4-methoxybenzyl)oxy)-6-(4(trifluoromethyl)phenyl)pyridine (17)

An argon-purged mixture of $\mathbf{1 6}(2.74 \mathrm{~g}, 6.52 \mathrm{mmol}, 1 \mathrm{eq}),. 4-$ (trifluoromethyl)phenylboronic acid ( $1.11 \mathrm{~g}, 5.87 \mathrm{mmol}, 0.9$ eq.), $\mathrm{Na}_{2} \mathrm{CO}_{3}$ ( $\left.1.38 \mathrm{~g}, 13.0 \mathrm{mmol}, 2 \mathrm{eq}.\right)$ and $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(377 \mathrm{mg}$, $0.33 \mathrm{mmol}, 0.05 \mathrm{eq}$.) in a mixture of toluene $/ \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(6 / 1 / 1)$ $(109 \mathrm{~mL})$ was stirred at $110^{\circ} \mathrm{C}$ for 15 h . The reaction mixture was cooled to $25^{\circ} \mathrm{C}$ and concentrated under reduced pressure. The residue was directly purified by flash chromatography on silica gel (cyclohexane/EtOAc 8/1) affording 17 ( $2.73 \mathrm{~g}, 95 \%$ ) as a yellow oil. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta \mathrm{H} 3.83(\mathrm{~s}, 3 \mathrm{H}), 5.08$ (s, 2H) , $6.95(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.07(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.27$
$(\mathrm{d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.36(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.70(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $2 \mathrm{H}), 8.04(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta \mathrm{C} 55.3,70.5,108.1,112.7,114.3,124.0(\mathrm{q}, J=272.2 \mathrm{~Hz}), 125.6$ (q, $J=3.8 \mathrm{~Hz}), 126.8,127.2,129.5,131.3(\mathrm{q}, J=32.6 \mathrm{~Hz})$, $141.0,143.1,157.4,160.0,166.6 \mathrm{ppm}$; IR $v_{\max }: 2935,1580,1321$ $\mathrm{cm}^{-1} ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}: 439[\mathrm{M}+\mathrm{H}]^{+}$.
4.1.12. 2-Iodo-4-((4-methoxybenzyl)oxy)-6-(4(trifluoromethyl) phenyl)pyridine (18)

To a solution of 17 ( $453 \mathrm{mg}, 1.03 \mathrm{mmol}, 1 \mathrm{eq}.), \mathrm{CuI}(10.0 \mathrm{mg}$, $0.05 \mathrm{mmol}, 0.05 \mathrm{eq}$.) and $\mathrm{NaI}(309 \mathrm{mg}, 2.06 \mathrm{mmol}, 2 \mathrm{eq}$.) in anhydrous 1,4 -dioxane $(1.3 \mathrm{~mL})$ were added $N, N^{\prime}$ dimethylethylenediamine $(11.0 \mu \mathrm{~L}, 0.10 \mathrm{mmol}, 0.1 \mathrm{eq}$.). The solution was stirred at $110^{\circ} \mathrm{C}$ for 28 h , cooled to $25^{\circ} \mathrm{C}$, treated with $\mathrm{NH}_{4} \mathrm{OH}\left(28 \%\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)(10 \mathrm{~mL})$ and diluted with water. The mixture was extracted with DCM. The organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 8/1) affording $18(442 \mathrm{mg}, 88 \%)$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta \mathrm{H} 3.74(\mathrm{~s}, 3 \mathrm{H}), 4.96(\mathrm{~s}, 2 \mathrm{H}), 6.86(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H})$, $7.17(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.21(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.26(\mathrm{~d}, J=$ $8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.59(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.93(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H})$ ppm; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{C} 55.3,70.3,108.4,114.2$, $118.8,119.7,124.0(\mathrm{q}, \quad J=272.3 \mathrm{~Hz}), 125.6(\mathrm{q}, J=3.7 \mathrm{~Hz})$, $126.9,127.2,129.5,131.2(\mathrm{q}, J=32.5 \mathrm{~Hz}), 141.0,157.9,160.0$, $165.5 \mathrm{ppm} ;$ IR $\nu_{\max }: 2935,1575,1319,1110,993,827 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{~F}_{3} \mathrm{BrINO}_{2}[\mathrm{M}+\mathrm{H}]^{+}$486.0178, found 486.0190.
4.1.13.4-((4-Methoxybenzyl)oxy)-2-(trifluoro methyl)-6-(4-(trifluoromethyl)phenyl)pyridine (19)

To a solution, under argon, of $18(2.28 \mathrm{~g}, 4.70 \mathrm{mmol}, 1 \mathrm{eq}$.$) ,$ $\mathrm{KF}(1.64 \mathrm{~g}, 28.2 \mathrm{mmol}, 6 \mathrm{eq}),. \mathrm{CuI}(0.36 \mathrm{~g}, 1.88 \mathrm{mmol}, 0.4 \mathrm{eq}$. and 1,10 -phenantroline $(0.34 \mathrm{~g}, 1.88 \mathrm{mmol}, 0.4$ eq.) in anhydrous DMSO $(68 \mathrm{~mL})$ were added $\mathrm{B}(\mathrm{OMe})_{3}(3.14 \mathrm{~mL}, 28.2 \mathrm{mmol}$, 6 eq.) and $\mathrm{TMSCF}_{3}(2 \mathrm{M}$ in THF) ( $14.1 \mathrm{~mL}, 28.2 \mathrm{mmol}, 6$ eq.). The solution was stirred at $60^{\circ} \mathrm{C}$ for 24 h . The reaction mixture was treated with $\mathrm{NH}_{4} \mathrm{OH}\left(28 \%\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)(20 \mathrm{~mL})$ and extracted with diethyl ether. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 4/1) affording 19 (1.78 g, 89\%) as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H} 3.75(\mathrm{~s}, 3 \mathrm{H}), 5.07(\mathrm{~s}, 2 \mathrm{H}), 6.88(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.17(\mathrm{~d}, J$ $=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.63(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.02(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{C} 55.3,70.5,106.6(\mathrm{q}, J=2.8 \mathrm{~Hz}), 109.7$, $114.3,121.4(\mathrm{q}, J=274.6 \mathrm{~Hz}), 124.0(\mathrm{q}, J=272.2 \mathrm{~Hz}), 125.7(\mathrm{q}$, $J=3.8 \mathrm{~Hz}), 126.7,127.4,129.5,131.5(\mathrm{q}, J=32.6 \mathrm{~Hz}), 141.2$, $150.0(\mathrm{q}, J=34.5 \mathrm{~Hz}), 157.9,160.0,166.7 \mathrm{ppm}$; IR $v_{\max }: 2935$, 1604, 1324, $1126 \mathrm{~cm}^{-1}$; MS (ESI ${ }^{+}$) m/z: $428[\mathrm{M}+\mathrm{H}]^{+}$.
4.1.14.2-(Trifluoromethyl)-6-(4-trifluoromethyl) phenyl)pyridin-4-ol (20)

To a solution of $19(1.78 \mathrm{~g}, 4.17 \mathrm{mmol}, 1 \mathrm{eq}$.$) in \mathrm{DCM}$ $(42 \mathrm{~mL})$ were added TFA $(1 \mathrm{~mL})$. The reaction mixture was stirred at $25^{\circ} \mathrm{C}$ for 7 h and treated with 1 M NaOH aq. ( $3 \times 20 \mathrm{~mL}$ ). The aqueous layers were acidified with 1 M HCl until $\mathrm{pH} \sim 6$ and concentrated under reduced pressure. The residue was washed with MeOH and the solid filtered. The filtrate was concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 1/1) affording $20(1.03 \mathrm{~g}, 80 \%)$ as a white solid. m.p.: $170^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta \mathrm{H} 7.14(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.48$ $(\mathrm{d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.77(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.20(\mathrm{~d}, J=8.2 \mathrm{~Hz}$,
$2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $8 \mathrm{C} 108.8(\mathrm{q}, \quad J=$ $2.8 \mathrm{~Hz}), 111.7,123.0(\mathrm{q}, J=273.5 \mathrm{~Hz}), 125.7(\mathrm{q}, J=271.3 \mathrm{~Hz})$, $126.7(\mathrm{q}, J=3.8 \mathrm{~Hz}), 128.7,132.4(\mathrm{q}, J=32.2 \mathrm{~Hz}), 143.0,150.8$ ( $\mathrm{q}, J=34.2 \mathrm{~Hz}$ ), $159.2,168.1 \mathrm{ppm}$; IR $v_{\max }: 3097,1611,1582$, 1323, 1108, 1061, $849 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{13} \mathrm{H}_{8} \mathrm{~F}_{6} \mathrm{NO}$ $[\mathrm{M}+\mathrm{H}]^{+} 308.0510$, found 308.0523 .

### 4.1.15.4-Bromo-2-(trifluoromethyl)-6-(4-(trifluoro methyl)phenyl)pyridine (21)

To a solution, under argon, of $20(6.5 \mathrm{~g}, 21.2 \mathrm{mmol}, 1 \mathrm{eq}$.$) in$ DMF ( 130 mL ) were added $\mathrm{POBr}_{3}(24.3 \mathrm{~g}, 84.6 \mathrm{mmol}, 4 \mathrm{eq}$.$) .$ The suspension was stirred at $110{ }^{\circ} \mathrm{C}$ for 16 h and quenched with $\mathrm{H}_{2} \mathrm{O}$. The resulting mixture was extracted with EtOAc. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 9/1) affording $21(6.02 \mathrm{~g}, 93 \%)$ as a white solid. m.p.: $59{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H} 7.76\left(\mathrm{~d},{ }^{3} \mathrm{~J}=\right.$ $8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.83(\mathrm{~s}, 1 \mathrm{H}), 8.12(\mathrm{~s}, 1 \mathrm{H}), 8.16\left(\mathrm{~d},{ }^{3} J=8.2 \mathrm{~Hz}, 2 \mathrm{H}\right)$ ppm; ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{C} 120.7\left(\mathrm{q},{ }^{1} \mathrm{~J}=275.1 \mathrm{~Hz}\right)$, $122.9\left(\mathrm{q},{ }^{3} J=2.8 \mathrm{~Hz}\right), 123.9\left(\mathrm{q},{ }^{1} J=272.4 \mathrm{~Hz}\right), 126.0\left(\mathrm{q},{ }^{3} J=\right.$ $3.8 \mathrm{~Hz}), 126.4,127.6,132.2\left(\mathrm{q},{ }^{2} J=32.7 \mathrm{~Hz}\right), 134.9$, 139.7, $149.3\left(\mathrm{q},{ }^{2} J=35.4 \mathrm{~Hz}\right), 157.5 \mathrm{ppm}$; IR $v_{\max }: 1573,1320,1119$, 1057, $843 \mathrm{~cm}^{-1}$..

### 4.1.16. (S)-1-(2-(Trifluoromethyl)-6-(4-(trifluoro methyl)phenyl)pyridin-4-yl)ethane-1,2-diol ((S)-22)

To a suspension of AD-mix $\alpha(1.26 \mathrm{~g})$ in a $t \mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}$ mixture $(1 / 1)(22 \mathrm{~mL})$ was added potassium osmate $(3.00 \mathrm{mg}$, $0.09 \mathrm{mmol}, 0.01 \mathrm{eq}$.$) . The mixture was cooled to 0^{\circ} \mathrm{C}$ and 4 ( $285.0 \mathrm{mg}, 0.90 \mathrm{mmol}, 1$ eq.) was added. The reaction mixture was stirred at $0^{\circ} \mathrm{C}$ for 5 min and at $25^{\circ} \mathrm{C}$ for 15 h . The reaction was cooled to $0^{\circ} \mathrm{C}$ and quenched with $\mathrm{Na}_{2} \mathrm{SO}_{3}(980 \mathrm{mg})$. The suspension was stirred for 10 min at $0^{\circ} \mathrm{C}, 10 \mathrm{~min}$ at $25^{\circ} \mathrm{C}$ and extracted with EtOAc. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was precipitated with cyclohexane. The resulting solid was filtered affording (S)-22 (316 mg, 99\%) as a white solid. m.p. $101{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta \mathrm{H} 3.77$ (m, $2 \mathrm{H}), 4.89(\mathrm{~m}, 1 \mathrm{H}), 7.76(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.84(\mathrm{~s}, 1 \mathrm{H}), 8.20(\mathrm{~s}$, $1 \mathrm{H}), 8.29(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $(100 \mathrm{MHz}$, $\left.\mathrm{CD}_{3} \mathrm{OD}\right): \delta \mathrm{C} 67.8,74.2,118.7(\mathrm{q}, J=2.8 \mathrm{~Hz}), 122.5,123.2(\mathrm{q}, J$ $=273.6 \mathrm{~Hz}), 125.6(\mathrm{q}, J=271.3 \mathrm{~Hz}), 126.8(\mathrm{q}, J=3.8 \mathrm{~Hz})$, $128.7,132.5(\mathrm{q}, J=32.3 \mathrm{~Hz}), 142.7,149.2(\mathrm{q}, J=34.3 \mathrm{~Hz})$, $157.0,157.2 \mathrm{ppm}$; IR $v_{\max }: 3293,1323,1097 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{~F}_{6} \mathrm{NO}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 374.0578$, found 374.0583; $[\alpha]_{\mathrm{D}}^{20}$ : $+29.1^{\circ}$ (с 0.1 ; MeOH); Chiral HPLC $99 \%$ ee, Chiralpak IB column, heptane $/ i-\mathrm{PrOH}, \quad 90 / 10$, flow $1 \mathrm{~mL} / \mathrm{min}, \quad \operatorname{tr}(R)=$ $10.6 \mathrm{~min}, \operatorname{tr}(S)=11.6 \mathrm{~min}$.
4.1.17. (R)-1-(2-(Trifluoromethyl)-6-(4-(trifluoro methyl)phenyl)pyridin-4-yl)ethane-1,2-diol ((R)-22)

To a suspension of AD-mix $\beta(8.85 \mathrm{~g})$ in a $t \mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}$ mixture $(1 / 1)(160 \mathrm{~mL})$ was added potassium osmate $(23.2 \mathrm{mg}$, $0.06 \mathrm{mmol}, 0.01 \mathrm{eq}$.$) . The mixture was cooled to 0^{\circ} \mathrm{C}$ and 4 $(2.00 \mathrm{~g}, 6.30 \mathrm{mmol}, 1$ eq.) was added. The reaction mixture was stirred at $0^{\circ} \mathrm{C}$ for 5 min and at $25^{\circ} \mathrm{C}$ for 15 h . The reaction was cooled to $0^{\circ} \mathrm{C}$ and quenched with $\mathrm{Na}_{2} \mathrm{SO}_{3}(7 \mathrm{~g})$. The suspension was stirred for 10 min at $0^{\circ} \mathrm{C}, 10 \mathrm{~min}$ at $25^{\circ} \mathrm{C}$ and extracted with EtOAc. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was precipitated with cyclohexane. The resulting solid was filtered affording ( $\boldsymbol{R}) \mathbf{- 2 2}(2.21 \mathrm{~g}, 99 \%)$ as a white solid. m.p., ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR and IR spectra are the same that those of $(\boldsymbol{S})$ 22. HRMS calcd. for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{~F}_{6} \mathrm{NO}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 374.0578$, found 374.0583; $[\alpha]_{\mathrm{D}}^{20}:-29.3^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC $>98,5 \%$ ee
(incomplete separation), Chiralpak IB column, heptane $/ i-\mathrm{PrOH}$, $90 / 10$, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=11.0 \mathrm{~min}, \operatorname{tr}(S)=12.7 \mathrm{~min}$.
4.1.18. (S)-4-( Oxiran-2-yl)-2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridine (S)-3

To a solution, under argon, of $(\boldsymbol{S})-\mathbf{2 2}(1.20 \mathrm{~g}, 3.42 \mathrm{mmol}$, 1 eq.) in anhydrous DCM ( 35 mL ) were added trimethylorthoacetate ( $1.3 \mathrm{~mL}, 10.25 \mathrm{mmol}, 3$ eq.) and $p \mathrm{TsOH}$ ( $32.5 \mathrm{mg}, 0.17 \mathrm{mmol}, 0.05 \mathrm{eq}$.). The reaction mixture was stirred at $25^{\circ} \mathrm{C}$ for 30 min and then concentrated under reduced pressure. The residue was placed under argon, dissolved in anhydrous DCM ( 35 mL ) and cooled to $0^{\circ} \mathrm{C}$. $\mathrm{TMSBr}(1.35 \mathrm{~mL}$, $10.25 \mathrm{mmol}, 3$ eq.) was then added dropwise. The solution was stirred at $0{ }^{\circ} \mathrm{C}$ for 5 min , at $25^{\circ} \mathrm{C}$ for 30 min and concentrated under reduced pressure. The residue was placed under argon and dissolved in anhydrous $\mathrm{MeOH}(35 \mathrm{~mL}) . \mathrm{K}_{2} \mathrm{CO}_{3}(2.36 \mathrm{~g}$, $17.08 \mathrm{mmol}, 5 \mathrm{eq}$.) was then added. The suspension was stirred at $25^{\circ} \mathrm{C}$ for 1 h , quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}$ and extracted with DCM. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 6/1) affording (S)-3 (1.08 g, 95\%) as a white solid. m.p. $81{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H} 2.84$ (dd, $J=5.6,2.5,1 \mathrm{H}), 3.29(\mathrm{dd}, J=5.6,4.3,1 \mathrm{H}), 4.02(\mathrm{dd}, J=$ $4.3,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{~m}, 1 \mathrm{H}), 7.75 \mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.86(\mathrm{~m}$, $1 \mathrm{H}), 8.18(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{C}$ $48.1(\mathrm{q}, J=4.1 \mathrm{~Hz}), 50.9,121.9(\mathrm{q}, J=275.6 \mathrm{~Hz}), 123.1,124.0$ (q, $J=272.2 \mathrm{~Hz}), 125.9(\mathrm{q}, J=3.7 \mathrm{~Hz}), 127.3,131.6(\mathrm{q}, J=$ $32.5 \mathrm{~Hz}), 132.1,135.5,140.6,145.6(\mathrm{q}, J=34.6 \mathrm{~Hz}), 154.3 \mathrm{ppm}$; IR $v_{\text {max }}: 2928,1319,1105,1068 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{~F}_{6} \mathrm{NO}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{H}]^{+}$334.0667, found 334.0679.; [ $\left.\alpha\right]_{\mathrm{D}}^{20}$ : $+23.5^{\circ}$ (c 0,$\left.1 ; \mathrm{MeOH}\right)$.
4.1.19. (R)-4-(Oxiran-2-yl)-2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridine ( $\boldsymbol{R}$ )-3)

To a solution, under argon, of $(\boldsymbol{R})-\mathbf{2 2}(1.00 \mathrm{~g}, 2.85 \mathrm{mmol}$, 1 eq.) in anhydrous DCM ( 30 mL ) were added trimethylorthoacetate ( $1.10 \mathrm{~mL}, 8.50 \mathrm{mmol}, 3$ eq.) and $p \mathrm{TsOH}$ $(27.0 \mathrm{mg}, 0.14 \mathrm{mmol}, 0.05 \mathrm{eq}$.$) . The reaction mixture was stirred$ at $25^{\circ} \mathrm{C}$ for 30 min and then concentrated under reduced pressure. The residue was placed under argon, dissolved in anhydrous DCM ( 30 mL ) and cooled to $0^{\circ} \mathrm{C}$. TMSBr $(1.13 \mathrm{~mL}$, $8.54 \mathrm{mmol}, 3 \mathrm{eq}$.) was then added. The solution was stirred at $0^{\circ} \mathrm{C}$ for 5 min , at $25^{\circ} \mathrm{C}$ for 30 min and concentrated under reduced pressure. The residue was placed under argon and dissolved in anhydrous $\mathrm{MeOH}(30 \mathrm{~mL}) . \mathrm{K}_{2} \mathrm{CO}_{3}(1.97 \mathrm{~g}$, $14.25 \mathrm{mmol}, 5 \mathrm{eq}$.) was then added. The suspension was stirred at $25^{\circ} \mathrm{C}$ for 2 h , quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}$ and extracted with DCM. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (cyclohexane/EtOAc 6/1) affording ( $\boldsymbol{R}$ )-22 $(804 \mathrm{mg}, 85 \%)$ as a white solid. m.p. $81{ }^{\circ} \mathrm{C}$; NMR and HRMS data in accordance with ( $\boldsymbol{S}$ )-22; IR $v_{\text {max }}: 2928,1319,1105,1068 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{~F}_{6} \mathrm{NO}_{2} \mathrm{Na} \quad[\mathrm{M}+\mathrm{H}]^{+}$334.0667, found 334.0679. $[\alpha]_{D}^{20}:-20.8^{\circ}(\mathrm{c} 0,1 ; \mathrm{MeOH})$.
4.1.20. General Procedure for the ring-opening reactions: Preparation of 4-AHPs (2)

To a solution of the epoxide $\mathbf{3}$ in ethanol $(0.05 \mathrm{M})$ was added the desired amine. The reaction mixture was stirred at $130^{\circ} \mathrm{C}$ using microwaves for 30 min and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel affording the corresponding 4-AHP.
4.1.21. (S)-2-(Butylamino)-1-(2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ((S)-2a)

Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{S}$ )-3 ( $200.0 \mathrm{mg}, 0.60 \mathrm{mmol}, 1 \mathrm{eq}$.) and $n$ butylamine ( $0.30 \mathrm{~mL}, 3.00 \mathrm{mmol}, 5.0 \mathrm{eq}$. ), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{S}$ )-2a ( $191.4 \mathrm{mg}, 78 \%$ ) as a white solid. m.p. $129^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta \mathrm{H} 0.93(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.31-$ $1.44(\mathrm{~m}, 2 \mathrm{H}), 1.45-1.56(\mathrm{~m}, 2 \mathrm{H}), 2.61-2.79(\mathrm{~m}, 3 \mathrm{H}), 3.06(\mathrm{dd}, J=$ $12.3,3.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.07\left(\mathrm{~s}_{\mathrm{br}}, 2 \mathrm{H}\right), 4.84(\mathrm{dd}, J=9.2,3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $7.67(\mathrm{~s}, 1 \mathrm{H}), 7.73(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.99(\mathrm{~s}, 1 \mathrm{H}), 8.18(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) : $\delta \mathrm{C} 13.9,20.2$, $32.0,49.0,56.2,69.7,116.6(\mathrm{q}, J=2.6 \mathrm{~Hz}), 120.0,124.0(\mathrm{q}, J=$ $272.3 \mathrm{~Hz}), 124.2(\mathrm{q}, J=274.8 \mathrm{~Hz}), 125.6(\mathrm{q}, J=3.8 \mathrm{~Hz}), 127.5$, $131.5(\mathrm{q}, J=32.6 \mathrm{~Hz}), 141.1,148.5(\mathrm{q}, J=34.6 \mathrm{~Hz}), 155.0$, 156.4 ppm ; IR $v_{\text {max }}: 2921,1328,1109 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+} 407.1558$, found $407.1551 ;[\alpha]_{\mathrm{D}}^{20}:+31.1^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC 99\% ee, Chiralpak IB column, heptane $/ i-\mathrm{PrOH} / \mathrm{EDA}, 99 / 1 / 0.1$, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=$ $17.8 \mathrm{~min}, \operatorname{tr}(S)=20.8 \mathrm{~min}$.
4.1.22. (R)-2-(Butylamino)-1-(2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ( $(\boldsymbol{R})-2 a)$

Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{R}) \mathbf{- 3}(40.0 \mathrm{mg}, 0.12 \mathrm{mmol}, 1 \mathrm{eq}$.$) and n$ butylamine ( $0.071 \mathrm{~mL}, 0.72 \mathrm{mmol}, 6 \mathrm{eq}$. ), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{R}$ )-2a ( $47 \mathrm{mg}, 96 \%$ ) as a white solid. m.p., ${ }^{1}$ H NMR, ${ }^{13} \mathrm{C}$ NMR and IR spectra are the same that those of (S)-2a. HRMS calcd. for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O} \quad[\mathrm{M}+\mathrm{H}]^{+} 407.1558$, found 407.1551; [ $\alpha]_{\mathrm{D}}^{20}:-31.0^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC $99 \%$ ee, Chiralpak IB column, heptane/i-PrOH/EDA, 99/1/0.1, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=17.7 \mathrm{~min}, \operatorname{tr}(S)=21.2 \mathrm{~min}$.
4.1.23. (S)-2-(Pentylamino)-1-(2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ( $(S)-2 b)$

Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{S}$ )-3 ( $200.0 \mathrm{mg}, 0.60 \mathrm{mmol}, 1 \mathrm{eq}$.) and $n$ pentylamine ( $0.36 \mathrm{~mL}, 3.10 \mathrm{mmol}, 5.15 \mathrm{eq}$. ), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{S}$ )-2b ( $236.6 \mathrm{mg}, 94 \%$ ) as a white solid. m.p. $112{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H} 0.89(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.20-$ $1.42(\mathrm{~m}, 4 \mathrm{H}), 1.52-1.62(\mathrm{~m}, 2 \mathrm{H}), 2.60-2.81(\mathrm{~m}, 3 \mathrm{H}), 3.09(\mathrm{dd}, J=$ $12.2,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.87\left(\mathrm{~s}_{\mathrm{br}}, 2 \mathrm{H}\right), 4.94(\mathrm{dd}, J=9.2,3.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.68(\mathrm{~s}, 1 \mathrm{H}), 7.73(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.00(\mathrm{~s}, 1 \mathrm{H}), 8.18(\mathrm{~d}, J=$ $8.2 \mathrm{~Hz}, 2 \mathrm{H}$ ) ppm; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{C} 13.9,22.5$, 29.3, 29.6, 49.3, 56.2, 69.7, 116.6 (q, $J=2.5 \mathrm{~Hz}$ ), 120.0, 124.0 $(\mathrm{q}, J=272.2 \mathrm{~Hz}), 124.2(\mathrm{q}, J=274.5 \mathrm{~Hz}), 125.7(\mathrm{q}, J=3.7 \mathrm{~Hz})$, $127.5,131.5(\mathrm{q}, J=32.5 \mathrm{~Hz}), 141.1,148.5(\mathrm{q}, J=34.6 \mathrm{~Hz})$, 155.0, 156.4 ppm ; IR $v_{\text {max }}$ : 2930, 1327, $1126 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+}$421.1715, found 421.1720; $[\alpha]_{D}^{20}$ : $+23.7^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC $99 \%$ ee, Chiralpak IB column, heptane $/ i-\operatorname{PrOH} / \mathrm{EDA}, 99 / 1 / 0.1$, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=$ $12.4 \mathrm{~min}, \operatorname{tr}(S)=14.3 \mathrm{~min}$.

### 4.1.24. (R)-2-(Pentylamino)-1-(2-(trifluoromethyl)-

 6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ( $(\boldsymbol{R})-2 b)$Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{R}$ ) $\mathbf{3}$ ( $200.0 \mathrm{mg}, 0.60 \mathrm{mmol}, 1 \mathrm{eq}$.) and $n$ pentylamine ( $0.36 \mathrm{~mL}, 3.10 \mathrm{mmol}, 5.15 \mathrm{eq}$. ), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{R})-\mathbf{2 b} \quad(214.0 \mathrm{mg}, 85 \%)$ as a white solid. m.p., ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR and IR spectra are the same that those of $(\boldsymbol{S})$ -

2b. HRMS calcd. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+}$421.1715, found 421.1720; $[\alpha]_{\mathrm{D}}^{20}:-27.9^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC $99 \% e e$, Chiralpak IB column, heptane/i-PrOH/EDA, 99/1/0.1, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=11.9 \mathrm{~min}, \operatorname{tr}(S)=14.0 \mathrm{~min}$.
4.1.25. (S)-2-(Hexylamino)-1-(2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ( $(S)-2 c$ )

Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{S}$ ) $\mathbf{- 3}(40.0 \mathrm{mg}, 0.12 \mathrm{mmol}, 1 \mathrm{eq}$.) and $n-$ hexylamine ( $0.05 \mathrm{~mL}, 0.36 \mathrm{mmol}, 3 \mathrm{eq}$.), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{S}$ )-2c ( $35 \mathrm{mg}, 67 \%$ ) as a white solid. m.p. $84{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H} 0.86(\mathrm{t}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.21-$ $1.37(\mathrm{~m}, 6 \mathrm{H}), 1.53-1.61(\mathrm{~m}, 2 \mathrm{H}), 2.63-2.82(\mathrm{~m}, 3 \mathrm{H}), 3.11(\mathrm{dd}, J=$ $12.2,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.76\left(\mathrm{~s}_{\mathrm{br}}, 2 \mathrm{H}\right), 5.00(\mathrm{dd}, J=9.4,3.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.67(\mathrm{~s}, 1 \mathrm{H}), 7.71(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.99(\mathrm{~s}, 1 \mathrm{H}), 8.16(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{C} 13.9,22.5$, 26.7, 29.4, 31.6, 49.3, 56.0, 69.5, 116.5 (q, $J=2.6 \mathrm{~Hz}$ ), 120.0, $124.0(\mathrm{q}, J=272.2 \mathrm{~Hz}), 124.2(\mathrm{q}, J=274.6 \mathrm{~Hz}), 125.7(\mathrm{q}, J=$ $3.8 \mathrm{~Hz}), 127.5,131.6(\mathrm{q}, J=32.6 \mathrm{~Hz}), 141.0,148.5(\mathrm{q}, J=$ $34.6 \mathrm{~Hz}), 154.7,156.4 \mathrm{ppm}$; IR $v_{\text {max }}: 2925,1325,1112 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+}$435.1871, found 435.1857; $[\alpha]_{\mathrm{D}}^{20}:+30.8^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC $99 \% e e$, Chiralpak IB column, heptane/i-PrOH/EDA, 99/1/0.1, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=9.7 \mathrm{~min}, \operatorname{tr}(S)=10.8 \mathrm{~min}$.
4.1.26. (R)-2-(Hexylamino)-1-(2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ( $(\boldsymbol{R})-2 c)$

Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{R}$ )-3 ( $50.0 \mathrm{mg}, 0.15 \mathrm{mmol}, 1 \mathrm{eq}$.) and $n$ hexylamine ( $0.06 \mathrm{~mL}, 0.45 \mathrm{mmol}, 3 \mathrm{eq}$.), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{R}$ )-2c ( $42 \mathrm{mg}, 65 \%$ ) as a white solid. m.p., ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR and IR spectra are the same that those of $(\boldsymbol{S})$-2c. HRMS calcd. for $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O} \quad[\mathrm{M}+\mathrm{H}]^{+}$435.1871, found 435.1857; $[\alpha]_{\mathrm{D}}^{20}:-28.6^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC $98 \% e e$, Chiralpak IB column, heptane $/ i-\mathrm{PrOH} / E D A, ~ 99 / 1 / 0.1$, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=9.6 \mathrm{~min}, \operatorname{tr}(S)=10.7 \mathrm{~min}$.
4.1.27. (S)-2-(Heptylamino)-1-(2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ( $(S)-2 d)$

Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{S}$ )-3 ( $50.0 \mathrm{mg}, 0.15 \mathrm{mmol}, 1 \mathrm{eq}$.) and $n$ heptylamine ( $0.13 \mathrm{~mL}, 0.90 \mathrm{mmol}, 6 \mathrm{eq}$. ), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{S}$ )-2d ( $40 \mathrm{mg}, 60 \%$ ) as a white solid. m.p. $76{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta \mathrm{H}$ 0.78-0.97 (m, 3H), 1.06-1.43 $(\mathrm{m}, 8 \mathrm{H}), 1.45-1.66(\mathrm{~m}, 2 \mathrm{H}), 2.62-2.70(\mathrm{~m}, 2 \mathrm{H}), 2.89[\mathrm{AB}(\mathrm{ABX})$, $J=12.2,8.6,2.7 \mathrm{~Hz}, 2 \mathrm{H}], 3.25-3.80\left(\mathrm{~s}_{\mathrm{br}}, 2 \mathrm{H}\right), 4.86[\mathrm{X}(\mathrm{ABX}), J=$ $8.6,2.7 \mathrm{~Hz}, 1 \mathrm{H}], 7.67(\mathrm{~s}, 1 \mathrm{H}), 7.74(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.00(\mathrm{~s}$, $1 \mathrm{H}), 8.19(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): §C 14.0, 22.6, 27.1, 29.1, 29.8, 31.7, 49.3, 56.1, 69.6, 116.6 (q, J $=2.6 \mathrm{~Hz}), 120.0,124.0(\mathrm{q}, J=272.2 \mathrm{~Hz}), 124.2(\mathrm{q}, J=$ $274.6 \mathrm{~Hz}), 125.8(\mathrm{q}, J=3.8 \mathrm{~Hz}), 127.5,131.2(\mathrm{q}, J=32.6 \mathrm{~Hz})$, $141.1,148.5(\mathrm{q}, J=34.7 \mathrm{~Hz}), 154.9,156.4 \mathrm{ppm}$; IR $v_{\text {max }}: 2928$, 1324, $1114 \mathrm{~cm}^{-1}$; HRMS calcd. for $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}]^{+}$ 449.2028, found 449.2040; [ $\alpha]_{D}^{20}:+24.9^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC 99\% ee, Chiralpak IB column, heptane $i$ - $\mathrm{PrOH} / E D A$, $99 / 1 / 0.1$, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=11.7 \mathrm{~min}, \operatorname{tr}(S)=13.5 \mathrm{~min}$.
4.1.28. (R)-2-(Heptylamino)-1-(2-(trifluoromethyl)-6-(4-(trifluoromethyl)phenyl)pyridin-4-yl)ethanol ( $\boldsymbol{R}$ )-2d)

Following the general method for the preparation of 4-AHPs and starting from ( $\boldsymbol{R}) \mathbf{- 3}(31.0 \mathrm{mg}, 0.09 \mathrm{mmol}, 1 \mathrm{eq}$.$) and n$ heptylamine ( $0.08 \mathrm{~mL}, 0.54 \mathrm{mmol}, 6 \mathrm{eq}$. ), the residue was purified by flash chromatography on silica gel (DCM/MeOH 9/1) affording ( $\boldsymbol{R}$ )-2d ( $34 \mathrm{mg}, 85 \%$ ) as a white solid. m.p., ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR and IR spectra are the same that those of (S)-2d. HRMS calcd. for $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O} \quad[\mathrm{M}+\mathrm{H}]^{+}$449.2028, found 449.2040; $[\alpha]_{\mathrm{D}}^{20}:-27.4^{\circ}$ (c 0.1 ; MeOH); Chiral HPLC $99 \% e e$, Chiralpak IB column, heptane/i-PrOH/EDA, 99/1/0.1, flow $1 \mathrm{~mL} / \mathrm{min}, \operatorname{tr}(R)=11.6 \mathrm{~min}, \operatorname{tr}(S)=13.5 \mathrm{~min}$.

### 4.2. Biological assay

The in vitro antiplasmodial activities were first tested over concentrations ranging from 39 nM to $40 \mu \mathrm{M}$ and then, if the molecule efficacy warranted it, further checked over a concentration range of 1 nM to $1 \mu \mathrm{M}$. The reference strains used were culture-adapted Plasmodium falciparum 3D7 and W2. The former strain is susceptible to chloroquine and displays a decreased susceptibility to mefloquine, while the latter is susceptible to mefloquine and resistant to chloroquine. Parasites were cultivated in RPMI medium (Sigma-Aldrich, Lyon, France) supplemented with 0.5\% Albumax I (Life Technologies corporation, Paisley, United Kingdom), hypoxanthine (SigmaAldrich), gentamicin (Sigma-Aldrich), and human erythrocytes. They were incubated at $37^{\circ} \mathrm{C}$ in a candle jar, as described previously. ${ }^{33}$ The $P$. falciparum drug susceptibility test was carried out in 96 -well flat bottom sterile plates under a final volume of $250 \mu \mathrm{~L}$. After 48-h incubation with the drugs, quantities of DNA in treated and control cultures of parasites in human erythrocytes were compared according to the SYBR Green I (Sigma-Aldrich) fluorescence-based method. ${ }^{33,35}$ Briefly, after incubation, plates were frozen at $-20^{\circ} \mathrm{C}$ until use. They were then left to thaw for 2 h at room temperature after which $100 \mu \mathrm{~L}$ of the homogenized culture were transferred to 96 -well flat bottom sterile black plates (Nunc Inc.) already containing $100 \mu \mathrm{~L}$ of the SYBR Green I lysis buffer ( $2 x$ SYBR Green, 20 mM Tris base $\mathrm{pH} 7.5,5 \mathrm{mM}$ EDTA, $0.008 \% \mathrm{w} / \mathrm{v}$ saponin, $0.08 \% \mathrm{w} / \mathrm{v}$ Triton X-100). A negative control, controls treated by solvents (DMSO and $\mathrm{H}_{2} \mathrm{O}$, typically), and positive controls (chloroquine and mefloquine) were added to each set of experiments. Plates were incubated for 1 h at room temperature and the SYBRGreen fluorescence was then read on a fluorescence plate reader (Tecan, Austria) using excitation and emission wavelengths of 485 and 535 nm , respectively. Concentrations inhibiting $50 \%$ of the parasite's growth (half maximal inhibitory concentration or $\mathrm{IC}_{50}$ values) were then calculated from the obtained experimental results using a regression program available on line. ${ }^{36}$

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## Supplementary Material

The supplementary material contains general procedure information, NMR spectra, HPLC analysis about the principal compounds and the X-ray analysis of (S)-3 and 5.

## Graphical Abstract

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[^0]:    *Corresponding author. Tel.: +33-(0)-322827494; fax: +33-(0)-322827469; e-mail:pascal.sonnet@u-picardie.fr

[^1]:    

    Racemic mixture of $(S, R)$ and $(R, S)$ $\mathrm{IC}_{50}($ PFD6 $)=34.4 \mathrm{nM} / \mathrm{IC}_{50}($ PAN2 $)=3.9 \mathrm{nM}^{6}$
    ${ }^{1 C_{50}}(S, R)<\mathrm{IC}_{50}(R, S): 1.7-1.8$ times $^{6}$ Toxicity of $(S, R) \ll(R, S)$

